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Energy

transformations – mostly involve heat and work

movements. The Fundamental law is the conservation of energy principle:

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energy cannot be  
created or  
destroyed, but can  
only be  
transformed from  
one form to  
another.

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design and direct

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the manufacturing,  
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first and second

law From the first

law:  $dU = dq + dW$

From the second

law:  $T dq \leq dS$

Where, for

irreversible system

$T dq \leq dS$  and, for

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reversible system

$dq = T ds$  For a closed system in which only

reversible pV work is involved  $dW = -pdV$  and  $T dq = T ds$

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thermodynamics),  
the 2nd law of  
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and the property relations. Energy can be viewed as the ability to cause changes. First law of

thermodynamics: one of the most fundamental laws of nature is the conservation of energy principle.

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